**Atoms, Bonds Review WS Name \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_**

1. Protons, neutrons, and electrons make up the atom
	1. Protons have a \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ charge
	2. Neutrons have a \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ charge
	3. Electrons have a \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ charge
	4. \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ and \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ are found in the nucleus of the atom.
	5. \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ are found orbiting the nucleus in shells.
2. What is the atomic mass and atomic number of Copper (Cu)?
	1. Atomic Mass = \_\_\_\_\_\_\_\_\_\_\_
	2. Atomic Number = \_\_\_\_\_\_\_\_\_
	3. Determine the number of each particle type in an atom of copper:

	Protons= \_\_\_\_\_\_\_
	Neutrons = \_\_\_\_\_\_\_\_\_
	Electrons = \_\_\_\_\_\_\_\_\_\_
3. Determine the number of each particle type in an atom of sodium (Na):
Protons= \_\_\_\_\_\_\_
Neutrons = \_\_\_\_\_\_\_\_\_
Electrons = \_\_\_\_\_\_\_\_\_\_
4. Complete the Bohr Diagram for an atom of Sodium



1. What are valence electrons?
2. Atoms bond in order to get a full valence shell. In general, how many valence electrons do atoms want to have? What is the exception?
3. There are two types of bonds: **ionic** and **covalent**.
 \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ bonding involves the transfer of electrons whereas
 \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ bonding involves the sharing of electrons.
4. What is the HONC Rule?